Sonochemical Synthesis of Hydrophilic Drug Loaded Multifunctional Bovine Serum Albumin Nanocapsules

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S Supporting Information

[AB](#page-6-0)STRACT: [A facile sono](#page-6-0)chemical approach is designed to fabricate protein nanocapsules for hydrophilic drugs (HDs), and HD-loaded multifunctional bovine serum albumin (BSA) nanocapsules (MBNCs) have been prepared for the first time. The as-synthesized HD-loaded MBNCs have a satisfying size range and an excellent magnetic responsive ability. Moreover, high-dose hydrophilic drugs could be loaded into the MBNCs. As carriers, HD-loaded MBNCs also show attractive redox-responsive controlled release ability for hydrophilic drugs and could be internalized selectively by the tumor cells through the folate-mediated endocytosis.

KEYWORDS: sonochemical, nanocapsule, hydrophilic drug, magnetic responsive, redox-responsive

1. INTRODUCTION

As a promising delivery system of nanometer or micrometer size ranges, protein nanocapsules and microcapsules have been a focus for many years, especially in the biological field and for the medical needs of biocompatible, biodegradable, and nontoxic materials.^{1−4} Certainly, many efforts have been propelled to design and prepare a variety of protein nanocapsules or microcapsules via nume[rou](#page-6-0)s techniques, such as chemical cross-linking, solvent evaporation method, microemulsion technology, microfluidic fabrication technique, layer-by-layer self-assembling, and so on. 5^{-8}

In recent years, sonochemistry has been developed rapidly as a novel preparation method, and sonochemical synthesis of pro[te](#page-6-0)i[n](#page-6-0) nanocapsules or microcapsules has drawn our eyes owing to its advantages as follows.⁹⁻¹¹ (i) The sonochemical synthesis is a simple, easy, and short-time physicochemical process dependent on the high-intensity [ul](#page-6-0)t[ras](#page-6-0)ound.¹² (ii) Through the sonochemical process, high dosage of hydrophobic drugs can be directly loaded into the protein nanoc[aps](#page-6-0)ules or microcapsules along with the water-insoluble solvent, which avoids destroying their structures.^{13−15} (iii) Although the fabrication of the protein nanocapsules and microcapsules is attributed to the sulfydryl cross-linki[ng](#page-6-0) [of](#page-6-0) the cysteine in the protein molecules, it does not affect the major α -helical structure and properties of protein molecules, for instance, biocompatibility and hydrophilicity, etc.¹⁶ (iv) The cross-linked disulfide bonds are easily broken in a reducing environment, so protein nanocapsules or microcapsu[les](#page-6-0) may control the release of hydrophobic drugs.^{17−19} (v) There are plenty of residual active groups (carboxyl, amino, etc.) in the as-synthesized protein nanocapsules or micr[ocaps](#page-6-0)ules; thus, various multifunctional protein nanocapsules or microcapsules can be fabricated further by modifying some functional groups or molecules.20,21 So far, there have been many reports about the sonochemical synthesis of the protein nanocapsules or microcapsules wit[h vari](#page-6-0)ous properties.¹

However, the sonochemical fabrication of the protein nanocapsules or microcapsules often occurs in the oil/water interface or gas/water interface;^{9,12} therefore, hydrophobic drugs dispersed in the oil phase can be easily encapsulated into the protein nanocapsules or microcap[sules](#page-6-0) while hydrophilic drugs (HDs) have no access. Hence, the development of the protein nanocapsules or microcapsules is restricted as drug carriers at some level due to their specialized loading for hydrophobic drugs.

Herein, a novel facile sonochemical fabrication of the protein nanocapsules is designed for the HDs for the first time, and HDs are loaded in multifunctional BSA nanocapsules (MBNCs) successfully. In the synthesis, folic acid (FA) and $Fe₃O₄$ magnetic nanoparticles $(Fe_3O_4 \text{ MNPs})^{22}$ are immobilized onto the BSA nanocapsules (BNCs) as targeted materials for the special delivery of the drugs, and me[an](#page-6-0)while hydrophilic rhodamine B (RhB) is encapsulated into the MBNCs as a model of the HDs. Through the subsequent analysis, the as-synthesized MBNCs show a good magnetic property, FA-mediated selectivity to tumor cells, and inherent redox-responsive ability. Hence, it is confidently believed that MBNCs will be good smart carriers for hydrophilic drugs.

2. EXPERIMENTAL SECTION

2.1. Materials. Ferrous chloride tetrahydrate $(FeCl, 4H, O, >99%)$ and ferric chloride hexahydrate (FeCl₃·6H₂O, >99%) were purchased from Tianjin Guangfu Chemical Reagents Company (Tianjin, China). Aqueous ammonia (NH4OH, 25%), N,N-dimethylformamide (DMF), and sodium citrate $(C_6H_5Na_3O_7, >99.0\%)$ were purchased from Beijing Chemical Reagent Company (Beijing, China). Span-80 was purchased from Aladdin Industrial Corporation (Shanghai, China). Bovine serum

Received: June 22, 2015 Accepted: August 14, 2015 Published: August 14, 2015 albumin (BSA), 1-ethyl-3-(3-dimethylaminepropyl) carbodiimide hydrochloride (EDC), and N-hydroxysuccinamide (NHS) were purchased from Shanghai Boao Biochemical Technology (Shanghai, China). Folic acid (FA), glutathione (GSH), gelatin, rhodamine B (RhB), and coumarin 6 (C6) were purchased from Sinopharm Chemical Reagent Limited Corporation (Shanghai, China). Phosphate buffers solution (PBS) was prepared by us. Other chemicals were of analytical grade and were used without further purification.

2.2. Synthesis of BSA@Fe₃O₄ MNPs. Fe₃O₄ MNPs with an average particle size of 20 nm were prepared through a chemical coprecipitation process.²³ Then, BSA (100 mg), EDC (100 mg), and NHS (70 mg) were dissolved in the PBS (pH 6.3), and the constant mechanical stirring was car[rie](#page-6-0)d out for 30 min to activate the carboxyl groups of BSA molecules. At the same time, the as-synthesized $Fe₃O₄$ MNPs (50 mg) was dispersed into the PBS (pH 6.3) under the ultrasound, forming Fe₃O₄ magnetic fluid. Afterward, Fe₃O₄ magnetic fluid was added to the BSA solution to be coated with BSA molecules. After fast stirring for 24 h in the 20 °C water bath, the core−shell BSA@Fe₃O₄ MNPs were collected and washed to remove the unreacted BSA molecules. Finally, BSA@Fe₃O₄ MNPs were dispersed into the deionized water.

2.3. FA Conjugated BSA@Fe₃O₄ MNPs. FA (5 mg) , NHS (42 mg), and EDC (60 mg) were dissolved in DMF/PBS solution $(v/v = 1:1, pH 6.3)$ by the stirring. Then, the solution with BSA@Fe₃O₄ MNPs (10 mg/mL) was added into the above-mentioned solution. After the reaction was carried out at room temperature for 24 h, FA functional BSA@Fe₃O₄ MNPs (FA-BSA@Fe₃O₄ MNPs) were collected using an external magnet. FA-BSA@Fe₃O₄ MNPs were dispersed in deionized water at last after being washed several times.

2.4. Preparation of HD-Loaded MBNCs. As a model HD, RhB was dissolved into gelatin solution (60 mg/mL) to make RhB-loaded hydrogel (1.0 mg/mL). Through the high-intensity ultrasound, RhBloaded hydrogel (1.0 mL) was dispersed into the vegetable oil (10 mL) with surfactant span-80 (0.5 mL) to form an RhB-loaded water/oil emulsion. The RhB-loaded water/oil emulsion (1 mL) was layered on top of the mixed solution (5 mL) containing FA-BSA@Fe₃O₄ MNPs $(50 \,\text{mg/mL})$ and BSA $(1 \,\text{mg/mL})$. The emulsion/water mixed solution was laid in the ice bath to maintain the temperature below 20 °C during the ultrasound treatment. Then the probe of an ultrasonicator was inserted into the emulsion/water mixed solution with its tip placed at the emulsion/water interface. An acoustic frequency at 20 kHz was carried out at a 400 W cm[−]² power for 5 min in the pulse sonication mode (2s/2s). After the sonication, the layered emulsion/water mixed solution turned into a brown suspension. The suspension was cooled naturally, and the resultant RhB-loaded MBNCs were separated from the suspension under the outer magnetic field and then washed several times and redispersed in deionized water (100 mg/mL).

2.5. Controlled Release of RhB-Loaded MBNCs. The assynthesized RhB-loaded MBNCs suspension (4 mL) was injected into dialysis tubing (8000). The dialysis tubing was immersed completely into the vessel with the PBS dialysate (pH 7.4, 100 mL) containing GSH (10 mM), and then the vessel was vibrated by 100 rpm in the water bath (37 °C). The absorbances of the dialysate at different time were measured by a UV−vis spectrophotometer at the characteristic absorption wavelength ($\lambda = 552$ nm), and the measurements were recorded. The absorbance of RhB in the dialysate was a direct diagnostic signal of the released amount of RhB-loaded MBNCs.

2.6. Cellular Uptake and Flow Cytometric Assay. For the cellular uptake of RhB-loaded MBNCs, tumor cells (A549 cells and HeLa cells) were seeded in 12-well plates where there were 1.0×10^5 cells per well and were incubated in serum-free medium for 24 h before the cellular endocytosis. Then the cells in some wells were incubated in serum-free medium with RhB-loaded MBNCs (2.0 mL) while the cells in other wells were still incubated in serum-free medium, and the cells in rest wells were pretreated with FA (10 mg/mL) for 1 h and then incubated in serum-free medium with RhB-loaded MBNCs (2 mL). After all the cells were incubated for another 6 h, they were harvested and washed with PBS (pH 7.4) and then fixed with ethanol (75%). Finally, the morphology of these fixed cells was characterized through the confocal laser scanning microscopy (CLSM).

For the flow cytometric assay, HeLa cells were seeded in 12-well plates where there were 2.0×10^5 cells per well and incubated in serumfree medium for some time. When the cells reached 70%−80% confluence, the cells in some wells were incubated in serum-free medium with RhB-loaded MBNCs (2.0 mL) for 6 h. In the meantime, the cells in other wells after the pretreatment with FA (1 h) were incubated in serum-free medium for 6 h. Subsequently, all the cells were harvested and suspended with PBS (pH 7.4), and then the flow cytometry was carried out using a FACS Calibur instrument.

2.7. Measurements. Transmission electron microscopy (TEM) images were recorded using the JEOL JEM-2100F transmission microscope at an accelerating voltage of 200 kV. The preparation samples dispersed in water were dropped onto an amorphous carbon-coated copper grid and allowed to air-dry at room temperature. The morphology of samples was investigated using the field emission scanning electron microscope (FESEM) with a Hitachi S-4800 FESEM. Confocal laser scanning microscopy (CLSM) images were taken with a confocal laser-scanning system TCLS attached to an inverse microscope (FV1000) from Olympus (Japan). The UV−vis absorption was observed by the UV-2550 spectrophotometer from Shimadzu (Japan). The size distribution of nanoparticles was measured by dynamic light scattering (DLS) with a 90Plus/BI-MAS. Fourier transform infrared spectra (FTIR) of KBr powder-pressed pellets were recorded in the range 400−4000 cm[−]¹ on a Nicolet Instruments Research Series 5PC Fourier transform infrared spectrometer. The flow cytometry was evaluated using a FACS Calibur instrument (BD Bioscience Mountain View). The microphotograph was obtained by an optical microscope. Thermogravimetric analysis (TG) was performed with a Pyris 1TGA (Perkin
Elmer) at a heating rate of 10 $^{\circ}\mathrm{C/min}$ in the nitrogen atmosphere over the range 100−900 °C.

3. RESULTS AND DISCUSSION

3.1. Fabrication of HD-Loaded MBNCs. According to the experiments, synthesis of HD-loaded MBNCs was a multistep process based on the sonochemical method (Scheme 1). First, the core−shell BSA@Fe3O4 MNPs were prepared by chemical

Scheme 1. Synthesis Mechanism of (A) FA-BSA@Fe₃O₄ MNPs, (B) HD-Loaded Micelles, and (C) HD-Loaded MBNCs

Figure 1. (A, C, and D) SEM images of HD-loaded MBNCs. (B) Size distribution of HD-loaded MBNCs.

bonding, and there were many sulfydryl groups remaining on the surface of the BSA@Fe₃O₄ nanoparticles.^{24,25} Then, BSA@ $Fe₃O₄$ nanoparticles were modified with the targeting molecule FA, which was confirmed by FTIR and UV−v[is spe](#page-6-0)ctra (Figure S1 and Figure S2). Second, HDs were dissolved in the gelatin hydrogel which was thermoresponsive (Figure S3), [and then](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf) [HD-loaded hyd](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf)rogel liquid was dispersed into the vegetable oil with span-80, forming a water/oil emulsi[on with a m](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf)ean size of 57 nm (Figure S4). Afterward, the emulsion cooled down to turn the inner hydrogel liquid into solid phase. Third, HD-loaded emulsio[n was laye](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf)red on the top of the mixed solution containing BSA and FA-BSA@Fe₃O₄ MNPs, and then a high-intensity ultrasound was carried out at the emulsion/water interface to fabricate HD-loaded MBNCs, in which BSA molecules and FA-BSA@Fe₃O₄ MNPs were cross-linked to form the shells of the nanocapsules with HD-loaded emulsion being coated as the cores.

3.2. Morphology of HD-Loaded MBNCs. As-synthesized HD-loaded MBNCs were dispersed well in the water, which was ascribed to hydrophilic BSA molecules of the capsule shells. Through the optical microscope, HD-loaded MBNCs in the water showed spherical morphology (Figure S5). SEM images could give us the superfine morphology of HD-loaded MBNCs. As shown in Figure 1A, HD-loaded [MBNCs](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf) were irregular spheres with size less than 1 μ m. The average diameter of HDloaded MBNCs was about 700 nm by the DLS with a 90Plus/ BI-MAS device (Figure 1B), and the size was enough to load the water/oil emulsion and to be relatively practicable for the pharmacological delivery in the bloodstream.^{26,27} Enlarging the

SEM images, we found most of the HD-loaded MBNCs were collapsed or folded due to the elimination of inner emulsion (Figure 1C), and it proved that HD-loaded MBNCs were soft material actually. When HD-loaded MBNCs were magnified further, the lumpy surfaces of HD-loaded MBNCs were observed clearly on account of the existence of $Fe₃O₄$ MNPs (Figure 1D).

The microscopic structure of HD-loaded MBNCs was measured by TEM as well. In Figure 2A, HD-loaded MBNCs with irregular morphology were thickly dotted with particle clusters on the capsule walls. T[hese parti](#page-3-0)cle clusters were found to be $Fe₃O₄$ MNPs after adjusting high magnification (Figure 2B). The case was in accord with the results of SEM. Furthermore, through the thermogravimetric analysis (TG), the [weight p](#page-3-0)roportion of $Fe₃O₄$ MNPs was about 12.85% (Figure S6). Because of strong superparamagnetism of $Fe₃O₄ MNPs₂^{28–31} HD-loaded$ MBNCs promised excellent magnetic respo[nsive prop](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf)erties.

3.3. Magnetic Property of HD-Loade[d](#page-7-0) [MB](#page-7-0)NCs. HDloaded MBNCs were demonstrated to have magnetic response by the collection and motion of HD-loaded MBNCs in an external magnetic field (Figure 3). When a magnet was placed on one side of the vial containing the brown HD-loaded MBNCs, HD-loaded MBNCs oc[curred to](#page-3-0) gather onto the side of the vial close to the magnet in a very short time, forming the aggregation (Figure 3A). After the removal of the magnet, it was easy to redisperse HD-loaded MBNCs into the solution by gentle s[haking. W](#page-3-0)hile the magnet was close to one side of the vial once more, HD-loaded MBNCs were aggregated again, demonstrating that HD-loaded MBNCs possessed satisfactory magnetic response. Moreover, HD-loaded MBNCs could be manipulated

Figure 2. TEM images of HD-loaded MBNCs (A and B).

Figure 3. (A) Collection and (B) motion of HD-loaded MBNCs.

to move in a desired direction by an external magnetic field; the location of HD-loaded MBNCs depended on the magnetic field (Figure 3B), and there was little magnetic loss with time. The phenomenon illustrated that HD-loaded MBNCs possessed a good magnetic targeted response which might increase the local drug concentration and improve the therapeutic efficiency. $32,33$

The magnetic property of HD-loaded MBNCs was also confirmed by the VSM measurement. The magnetization [curve](#page-7-0) of HD-loaded MBNCs in Figure 4 showed no detectable coercivity in the field sweep, implying that HD-loaded MBNCs were superparamagnetic. Hence, it could be concluded that the obtained HD-loaded MBNCs had excellent magnetic properties and had access to the field of targeted delivery.

Figure 4. Magnetization curves of FA-BSA@Fe₃O₄ MNPs and HDloaded MBNCs.

3.4. Loading Capacity of HD-Loaded MBNCs. To confirm whether HDs were encapsulated into the MBNCs, rhodamine B (RhB) , a widely used hydrophilic fluorescent dye, $34,35$ was loaded into MBNCs as a model HD. CLSM measurement was used to determine the RhB-loaded MBNCs, and the resu[lts w](#page-7-0)ere showed in Figure 5. The red signals could be seen clearly in the fluorescence mode (Figure 5B) while there were no signals disappearing in [the tran](#page-4-0)smission mode (Figure 5A). Obviously, the red fluorescent [signals](#page-4-0) were originating from RhB motivated by the laser. Superimposing [the two m](#page-4-0)odes (Figure 5C), we found that the red signals were from the RhB-loaded MBNCs. It demonstrated that the hydrophilic RhB w[as encap](#page-4-0)sulated successfully into the MBNCs.

In order to verify the loading mechanism of RhB further, hydrophobic fluorescent dye coumarin 6 was dispersed into the water/oil emulsion and then loaded into the MBNCs. Images D−G in Figure 5 showed the CLSM images of the MBNCs with coumarin 6 and RhB, and both green signals of coumarin 6 and red sign[als of RhB](#page-4-0) were from the capsules, which proved clearly that RhB was loaded into the MBNCs along with the water/oil emulsion. Meanwhile, the HD-loaded capacity of the MBNCs was measured by a UV-vis spectrophotometer (Figure 6). Through the absorbance−concentration curve of RhB (Figure S7), we could calculate that there was up to 86.67% of RhB e[ncapsulate](#page-4-0)d into the MBNCs. Accordingly, the MBNCs coul[d be goo](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf)d carriers for the hydrophilic drugs, for example, hydroxyurea (HU), nimustine hydrochloride (ACNU), and cytosine arabinoside (Ara-C), etc.

3.5. Controlled Release. Since HD-loaded MBNCs were closely related to the formation of the disulfide linkage, 36 HD-loaded MBNCs would be disassembled and release inner hydrophilic drugs in the reductive condition (Figure 7A), such [as](#page-7-0)

Figure 5. CLSM images of RhB-loaded MBNCs: (A) transmission mode, (B) fluorescence mode, and (C) merged mode. CLSM images of the MBNCs with coumarin 6 and RhB: (D) transmission mode, (E) red fluorescence mode, (F) green fluorescence mode, and (G) merged mode of the red and green fluorescence.

Figure 6. (a) UV-vis spectrum curve of RhB solution (100 μ g/mL). (b) UV−vis spectrum curve of RhB released from the MBNCs (1 mL).

DL-dithiothreitol (DTT) and glutathione (GSH) ;³⁷⁻³⁹ that is, HD-loaded MBNCs possessed a redox-sensitive property for drug controlled release. To confirm the controlled [releas](#page-7-0)e ability of HD-loaded MBNCs, some GSH-triggered disassembly experiments were carried out for the RhB-loaded MBNCs. The RhB-loaded MBNCs were treated in GSH dialysate (10 mM); meanwhile, the release amount of RhB in the dialysate was recorded by the UV−vis spectrophotometer at the maximum absorption wavelength of RhB $(\lambda_{\text{max}} = 552 \text{ nm})$,^{34,40} and then the time-dependent release curve of the RhB was plotted (Figure 7B). The RhB-loaded water/oil emulsion almost rel[eased](#page-7-0) all the RhB easily. When RhB-loaded water/oil emulsion was e[ncapsulat](#page-5-0)ed into the MBNCs by the sonochemical method, the release rate of the RhB-loaded MBNCs decreased sharply, and the final release amount also dropped a lot. By contrast, the release process of the RhB-loaded MBNCs was accelerated obviously in the presence of GSH, and the final release amount was up to 91.42%. These results suggested that HD-loaded MBNCs had excellent controlled release ability for their redox-responsive ability.

3.6. Cellular Selectivity. RhB-loaded MBNCs had many FA molecules on the capsule walls, so we chose A549 cells (lung cancer cells) and HeLa cells (ovarian carcinomas cells) to

Figure 7. (A) Schematic of the GSH-triggered disassembly of HD-loaded MBNCs; (B) time-dependent release curves: (a) RhB-loaded water/oil emulsion, (b) RhB-loaded MBNCs, and (c) RhB-loaded MBNCs in the presence of GSH.

monitor the cellular internalizing behaviors of HDs-loaded MBNCs.41,42 CLSM images of tumor cells incubated in the serum-free medium with RhB-loaded MBNCs had been shown in Figur[e 8. In](#page-7-0) contrast to A549 cells with fewer folate receptors (FRs) (images A and B), FR-overexpressed HeLa cells had much

Figure 8. CLSM images of tumor cells incubated in the serum-free medium with RhB-loaded MBNCs: A549 cells (A and B), HeLa cells (C and D), and HeLa cells after the pretreatment of FA (E and F). Among these images, images a, c, and e were in the bright field while images b, d, and f were in the dark field.

stronger red fluorescence in the intracellular regions (images C and (D) ; $43,44$ however, after the pretreatment of FA, the fluorescent intensity of HeLa cells was weakened a lot (images E and F)[. The](#page-7-0)se results implied that RhB-loaded MBNCs could be internalized easily by the FR-overexpressed tumor cells.

The flow cytometry also gave us some insight into the mechanism for cellular internalization of RhB-loaded MBNCs. As Figure 9A showed, the intracellular fluorescent intensity of HeLa cells incubated in pure serum-free medium was very low (curve a). However, the fluorescent intensity of HeLa cells incubated in serum-free medium with RhB-loaded MBNCs was very high (curve c). After HeLa cells were pretreated with FA, the fluorescent intensity was degraded obviously (curve d). The results of the flow cytometric assay were consistent with those in the CLSM images in Figure 8. Thus, we confirmed adequately that RhB-loaded MBNCs could be taken in mainly via the FA-dependent endocytosis (Figure 9B).⁴⁵ Certainly, hydrophobic drugs could be loaded into the MBNCs with HDs and internalized by the tumor cells (Figure S8)[, p](#page-7-0)erhaps improving the pharmacological effects synergistically.

In fact, compared with nor[mal cells, t](http://pubs.acs.org/doi/suppl/10.1021/acsami.5b05558/suppl_file/am5b05558_si_001.pdf)here were abundant GSH species in the tumor cells.^{46,47} Once HDs-loaded MBNCs were taken into the tumor cells by the endocytosis, GSH in the cells could cause the disassem[bly o](#page-7-0)f HD-loaded MBNCs and release inner hydrophilic drugs, which had a chance to realize drug controlled release. The behavior might help to overcome

Figure 9.(A) Flow cytometry of HeLa cells incubated in different nutrient solutions: (a) pure serum-free medium, (b) pure serum-free medium after the pretreatment of FA, (c) serum-free medium with RhB-loaded MBNCs, and (d) serum-free medium with RhB-loaded MBNCs after the pretreatment of FA. (B) Schematic of cellular uptake of HD-loaded MBNCs.

the multidrug resistance (MDR) of cancer cells.^{48−50} Therefore, HD-loaded MBNCs had access to the physiological therapeutics as the drug carriers.

4. CONCLUSION

In summary, a facile approach based on the sonochemical technique was rationally designed to fabricate multifunctional BSA nanocapsules for hydrophilic drugs, and high-dose hydrophilic drugs were encapsulated successfully into the MBNCs. HD-loaded MBNCs had a satisfying size range and showed an excellent magnetic responsive ability. As the carriers, HD-loaded MBNCs showed attractive controlled release ability for HDs owing to their redox responsiveness and had access to the cellular uptake through the FA-dependent endocytosis. Accordingly, HDs-loaded MBNCs should be excellent candidates as smart carriers for the targeted delivery and controlled release of HDs. Furthermore, the facile sonochemical route provided a chance to solve the loading of hydrophilic drugs for the protein nanocapsules or microcapsules.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsami.5b05558.

Results from analytical DSC; microphotograph; TG, DLS, [FTIR, CLSM, UV](http://pubs.acs.org)−vis meas[urements; and camera ima](http://pubs.acs.org/doi/abs/10.1021/acsami.5b05558)ges (PDF)

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Notes

The aut[hors declare no com](mailto:cui_xj@jlu.edu.cn)peting financial interest.

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